

Roquefortine, an Intermediate in the Biosynthesis of Oxaline in Cultures of *Penicillium oxalicum*

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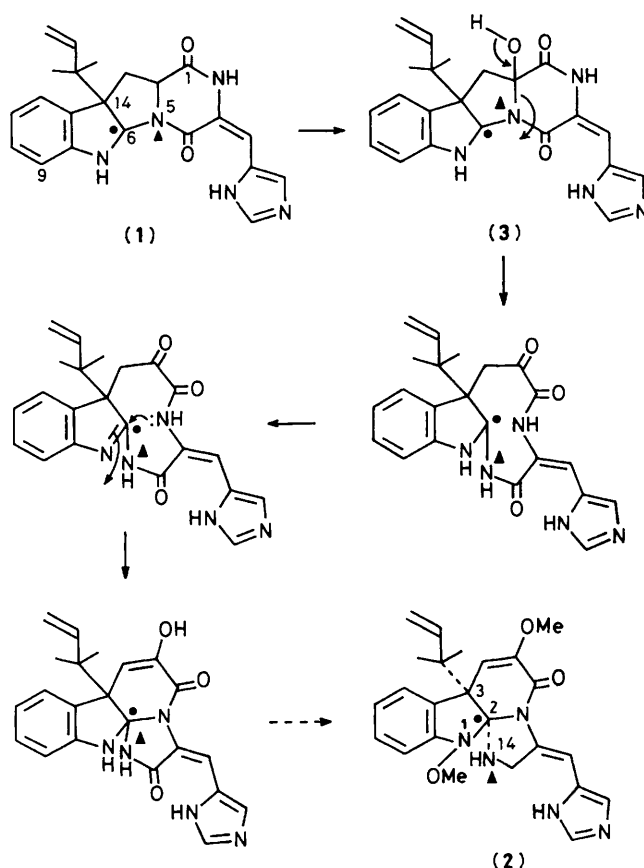
The incorporation of (2*RS*)-[*indole*-2-¹³C, 2-¹⁵N]tryptophan into both roquefortine and oxaline, and the efficient conversion of roquefortine into oxaline, defines a biosynthetic pathway by which N-14 of oxaline is derived from N-5 of roquefortine (and thus from tryptophan) and not from an exogenous nitrogen source.

Previous studies on the biosynthesis of roquefortine (1), a neurotoxic metabolite isolated from cultures of *Penicillium roqueforti*¹ and *Penicillium crustosum*² and oxaline (2), a metabolite of *Penicillium oxalicum*,^{3,4} established that the 3-*pro-S* hydrogen atom of (2*S*)-histidine is lost during the *in vivo* formation of the α,β -dehydroamino acid moiety in these two compounds.⁵ The mode of incorporation of the 1,1-dimethylallyl group at C-14 in roquefortine (1) has been studied with the aid of [1,2-¹³C₂]acetate and (3*RS*)-[2,3-¹³C₂]mevalonic acid lactone.⁶

A unique structural feature of oxaline (2) is the presence of a carbon atom, C-2, with three different nitrogen substituents. A plausible mechanism for the formation of this moiety requires that N-14 in oxaline originates from N-5 in roquefortine by an intramolecular rearrangement. The co-occurrence of roquefortine with oxaline in cultures of *P. oxalicum* lends credence to this hypothesis which was investigated using (¹⁴C)-, (³H,¹⁴C)-, and (¹³C,¹⁵N)-labelled precursors.

Ring closure of *N*-acetyl-(2*RS*)-tryptophan by dicyclohexylcarbodi-imide in pyridine gives the corresponding oxazolone which on addition of tritiated water undergoes rapid exchange at the chiral centre as well as slow hydrolytic opening of the oxazolone ring.^{7,8} Basic hydrolysis of the tritiated *N*-acetyltryptophan then afforded (2*RS*)-[2-³H]tryptophan. (2*RS*)-[*indole*-2-¹³C,2-¹⁵N]tryptophan was prepared from [2-¹³C]-indole-3-aldehyde and [1-¹⁵N]hydantoin⁹ (99.0 atom % ¹⁵N) using standard methods.¹⁰ The [2-¹³C]indole-3-aldehyde was in turn prepared from 2-nitrobenzaldehyde and potassium [¹³C]cyanide (90.0 atom % ¹³C) using reported procedures.¹¹

Incorporation of (2*S*)-[*ring*-2-¹⁴C]histidine (4.4%, dilution 15.1 †) and (2*S*)-[3-¹⁴C]tryptophan (10.8%, dilution 19.8) into



Scheme 1. Proposed biosynthetic pathway for oxaline (● = ¹³C, ▲ = ¹⁵N).

roquefortine by cultures of *P. roqueforti* (HPB 061175), on a yeast extract (2%)-sucrose (5%) medium confirmed the biosynthetic origin of the metabolite. Addition of a mixture of

† Dilution values obtained from ¹⁴C experiments are defined as in equation (1) where *n* and *m* are the appropriate number of

$$\text{Dilution} = \frac{\text{specific activity (precursor)}}{\text{specific activity (product)}} \times \frac{m(\text{product})}{n(\text{precursor})} \quad (1)$$

labelled sites; cf. P. S. Steyn, R. Vleggaar, and P. L. Wessels, *J. Chem. Soc., Perkin Trans. 1*, 1981, 1298.

(2*RS*)-[2-³H]- and (2*RS*)-[3-¹⁴C]-tryptophan (³H:¹⁴C 7.02) to cultures of *P. roqueforti* gave roquefortine with a ³H:¹⁴C ratio of 2.01 (incorporation of ¹⁴C 5.9%, dilution 12.0). This ratio corresponds to a 71.4% loss of ³H which can be ascribed to a combination of two factors *viz.* the conversion of (2*R*)-[2-³H]tryptophan *via* indole pyruvic acid into (2*S*)-tryptophan with concomitant loss of ³H and exchange of ³H at C-16 in roquefortine with ¹H of the medium. Thus in feeding experiments with [2-¹⁵N]tryptophan a substantial loss of ¹⁵N label can be expected.

The fate of the C-2 nitrogen atom of tryptophan in the biosynthesis of roquefortine was studied by the addition of (2*RS*)-[indole-2-¹³C, 2-¹⁵N]tryptophan, admixed with (2*RS*)-[benzene-ring-U-¹⁴C]tryptophan to provide an independent measure of dilution, to cultures of *P. roqueforti*. Incorporation of the ¹⁴C precursor into roquefortine was found to occur with high efficiency (6.1%) and low dilution (8.7). The presence of ¹³C at C-6 in the enriched roquefortine directly bonded to ¹⁵N was evident from the (¹³C, ¹⁵N)-coupling (¹J 7.0 Hz) observed for the C-6 resonance at δ 78.3 p.p.m. in the 20.0 MHz proton-noise-decoupled ¹³C n.m.r. spectrum. The enhanced intensity of the C-6 resonance is ascribed to the presence of (¹³C, ¹⁴N)-labelled roquefortine and indicated that *ca.* 64% of the ¹⁵N label of the tryptophan precursor is lost.

The origin of N-14 in oxaline was studied in the first instance by the addition of (2*RS*)-[indole-2-¹³C, 2-¹⁵N]tryptophan, admixed with (2*RS*)-[benzene-ring-U-¹⁴C]tryptophan, to cultures of *P. oxalicum* (MRC 100) on yeast extract (2%)-sucrose (5%) medium. The ¹⁴C precursor was incorporated efficiently (3.0%) with low dilution (20.5) into oxaline. The 125.76 MHz broad-band proton-decoupled ¹³C n.m.r. spectrum of the enriched oxaline pointed to the presence of contiguous ¹³C and ¹⁵N atoms as the resonance at δ 101.60 p.p.m., assigned to C-2, exhibited a one-bond (¹³C, ¹⁵N)-coupling of 10.1 Hz. In addition the enhanced intensity of this resonance, due to the presence of [2-¹³C, 14-¹⁴N]oxaline, showed that *ca.* 37% of the ¹⁵N label of the tryptophan precursor is lost. The precursor role of roquefortine in the biogenesis of oxaline by cultures of *P. oxalicum* was confirmed using the [benzene-ring-U-¹⁴C, 6-¹³C, 5-¹⁵N]roquefortine obtained from cultures of *P. crustosum* (Sol-7) supplemented with (2*RS*)-[indole-2-¹³C, 2-

¹⁵N]tryptophan (incorporation 4.3%, dilution 10.1). The roquefortine was efficiently incorporated into oxaline (24.3%) but the dilution of 29.8 precluded the observation, even at 125.76 MHz, of the one-bond (¹³C, ¹⁵N)-coupling for the C-2 resonance in the broad-band proton-decoupled ¹³C n.m.r. spectrum.

A plausible mechanism for the formation of oxaline from roquefortine is presented in Scheme 1. The isolation and characterization (by mass spectrometry and n.m.r. spectroscopy) of a new metabolite, 16-hydroxyroquefortine (3), C₂₂H₂₃N₅O₃, together with roquefortine from large-scale fermentations of *P. crustosum* (Sol-7) supports the proposed pathway.

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